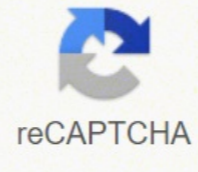




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**3.3 1948: The Revolution of the Liberals**

Subsequent to the 1947 constitutional conference, the Indian National Congress and the Indian National League (INL) were the two main political parties in India. The INL was formed in 1947 and was the main political party in the Indian subcontinent. The INL was formed in 1947 and was the main political party in the Indian subcontinent. The INL was formed in 1947 and was the main political party in the Indian subcontinent.

**Source C**

There were liberty and equality for women to be achieved.

The Indian National Congress was the main political party in India. The INL was formed in 1947 and was the main political party in the Indian subcontinent. The INL was formed in 1947 and was the main political party in the Indian subcontinent.

**Source D**

The Indian National Congress was the main political party in India. The INL was formed in 1947 and was the main political party in the Indian subcontinent. The INL was formed in 1947 and was the main political party in the Indian subcontinent.

**Source E**

The Indian National Congress was the main political party in India. The INL was formed in 1947 and was the main political party in the Indian subcontinent. The INL was formed in 1947 and was the main political party in the Indian subcontinent.

**The Rise of Nationalism in Europe**

In 1848, the Revolutions of 1848 were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe.

**Source A**

The Revolutions of 1848 were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe.

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The Revolutions of 1848 were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe.

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The Revolutions of 1848 were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe.

**Source D**

The Revolutions of 1848 were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe. The revolutions were a series of political revolutions that spread across Europe.

**Names given by Europeans to Countries of the 'New World'**

'AMERICA'	First used after the publication of the travels of Amerigo Vespucci (1451-1512)
'CANADA'	from <i>kanata</i> (= 'village' in the language of the Huron-Iroquois, as heard by the explorer Jacques Cartier in 1535)
'AUSTRALIA'	Sixteenth-century name for land in the Great Southern Ocean ( <i>australis</i> is Latin for 'south')
'NEW ZEALAND'	Name given by Tasman of Holland, who was the first to sight these islands in 1642 ( <i>zee</i> is Dutch for 'sea')

The *Geographical Dictionary* (pp 805-22) lists over a hundred place-names in the Americas and Australia which begin with 'New'.

The Aristocracy and the New Middle Class

socially and politically, a landed aristocracy was the dominant class in the continent. They owned estates and property both in countryside and town houses. In western and Central parts of Europe, there was an emergence of commercial classes whose existence was based on production for the market. Due to industrialisation, the new social groups - a working

**Table 1: Key Concepts**

1. The scientific method is a systematic approach to acquiring knowledge through observation and experimentation.
2. The scientific method involves the formulation of a hypothesis, the design of an experiment, the collection of data, and the analysis of the results.
3. The scientific method is a process that is used to test a hypothesis and to determine whether it is supported or refuted by the evidence.
4. The scientific method is a process that is used to test a hypothesis and to determine whether it is supported or refuted by the evidence.
5. The scientific method is a process that is used to test a hypothesis and to determine whether it is supported or refuted by the evidence.

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Gase characteristics: Gases do not have a defined volume or a defined form. MOLALITY 1. All non -zero days are significant. The chemical equation is (V) Avogadro Law: Avogadro indicates the position of decimal point. → KelvinĀĀA scale of temperature is S.I. scale and is very common these days. Temperature on this scale is shown by the sign K. Counting numbers of objects. The mass of a substance can be determined accurately in the laboratory by using an analytical balance. 7. Atoms of different elements differ in mass. In SI units, volume is expressed in metrc3 (m3). All the organic compounds have been found to contain carbon as their essential constituent. (D) Application in Industry: Chemistry has played an important role in developing many industrially c manufactured fertilizers, alkalis, acids, salts, dyes, polymers, drugs, soaps, detergents, metal alloys and other inorganic and organic chemicals including new materials contribute in a big way to the national economy. All of them have been fabricated. This reactant which reacts completely in the reaction is called the limiting reactant or limiting reagent. This ratio cannot change. Mole fraction: It is the ratio of number of moles of a particular component to the total number of moles of the solution. A few examples of mixtures are: milk, sea water, petrol, lime water, paint glass, cement, wood etc. The final result has two decimal places but the answer has to be reported only up to one decimal place, i.e., the answer would be 12.0. Subtraction of numbers can be done in the same way as the addition. Liquids: In liquids, the particles are close to each other but can move around. These are stated below: 1. The final result has four decimal places. Some examples of physical properties are colour, odour, melting point, etc. Similarly, the glucose formula is C6H12O6. But only up to two decimals should be reported, that is, the answer would be 11.36. Is he preparing for the exams? Study without Internet (outline) Share this with your friends Subscribe āĀc The importance of chemistry Chemistry has a direct impact on our lives and has a wide range of applications in different fields. (ii) Organic compounds are compounds that are present in plants and animals. (iii) The use of preservatives has helped



preserve food products such as jam, butter, pumpkins, etc. â€¢ Classification of matter There are two ways to classify the subject: (a) Physical classification (b) Chemical classification (a) Physics Classification: matter may exist in three physical states: 1. Gases: in gases, particles are very separate compared to those present in solid or liquid states. Therefore, hydrogen, nitrogen and oxygen gases consist of molecules in which two atoms combine to give the respective molecules of the element. For example, (iv) Lussac’s gaseous volume law, the law states that, in similar temperature and pressure conditions, whenever the gases are combined, they do so in volumes that have a simple integer ratio between themselves and also with the Gaseous Products. A balanced equation for this reaction is the following: limiting the reactive/reactive at times, in the alchemical equation, the present reagents are not the required amount according to the balanced equation. Today, dysentery and pneumonia are curable due to the discovery of sulphur drugs and drugs that save the life of penicillin. Kilogram: It is the mass unit. For example, the molecular mass of methane (CH4) = 12.011 U + 4 (1.008 U) = 16.043 U Ionic compounds of mass U such as NaCl, KNO3, NA2CO3, etc. For example: whenIron archives and sulfur powder are mixed, the mixture formed is heterogegâ € neo. Everyone contains types of a type. Weight: is the force exerted by gravity in object. â € ¢ Molecular mass: The molecular mass of a molecule is obtained by taking sum of atoms of different enthusiasts present in a molecule. For a solution containing Topos N2 of the solute dissolved in moles N1 of the solvent, 3. To express its empirical formula, we have to get a common factor 2. It is denoted by m. Can vary 3, 3.

Into Algebra 1, Geometry, Algebra 2, 8-12 Into Math, K-8 Math Expressions, PreK-6 Thank you for making Chowhound a vibrant and passionate community of food trailblazers for 25 years. We wish you all the best on your future culinary endeavors. A lecture (from Latin l ct ra “reading”) is an oral presentation intended to present information or teach people about a particular subject, for example by a university or college teacher.Lectures are used to convey critical information, history, background, theories, and equations. A politician’s speech, a minister’s sermon, or even a business person’s sales presentation may be similar ... Circles. The set of all the points in a plane that is at a fixed distance from a fixed point makes a circle. A Fixed point from which the set of points are at fixed distance is called the centre of the circle. A circle divides the plane into 3 parts: interior (inside the circle), the circle itself and exterior (outside the circle) For More Information On Terms Related To Circles, Watch The ... Ch-1. Number System. Contains seven exercises from 1A to 1G: 1A - This exercise is a reference for helping students know the methods to find rational numbers between two given numbers. 1B contains 5 Questions. 1C contains 14 questions. 1D contains 8 questions. 1E contains 10 questions. 1F contains 25 questions. 1G contains 18 questions ...

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